

Formulas and Names I

Name _____ Per ____

Write the Formula for the following ionic compounds.

1. Calcium + Chlorine _____
2. Sodium + Oxygen _____
3. Potassium + Sulfur _____
4. Calcium + Iodine _____
5. Magnesium + Nitrogen _____
6. Copper (II) + Bromine _____
7. Lead (II) + Chlorine _____
8. Cobalt (III) + Fluorine _____
9. Nickel (II) + Iodine _____
10. Iron (III) + Oxygen _____

Write the Names for the next set of ionic compounds.

11. Li_2O _____
12. BeCl_2 _____
13. KCl _____
14. Ag_2S _____
15. MgBr_2 _____

Write the Names for the next set of ionic compounds having multiple oxidation states.

16. Co_2O_3 _____
17. FeO _____
18. CrCl_3 _____
19. PbCl_2 _____
20. TiN _____